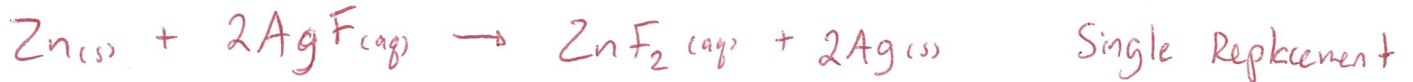


REDOX PRACTICE

1. For the following reaction:

- Identify the type of chemical reaction
- Predict and balance the chemical equation
- Write the net ionic equation (start with the total ionic equation)
- Label and write the balanced half-reactions
- Label the OA and RA

Zinc metal is combined with aqueous silver fluoride.



2. Write and label two balanced half-reaction equations to describe the reaction of nickel metal with aqueous lead (II) nitrate. Identify the OA and RA in the net REDOX equation.

